

Fluid Statics

- Fluids can NOT remain at rest under the presence of shear stress.
- In other words, fluids at rest can NOT support any shear.
- For static fluids we can only talk about normal stress which is equal to pressure.
- Determining the pressure distribution within a static fluid is the main task here.
- Applications include
- Pressure distribution in still atmosphere and oceans.
- Pressure measurement using manometers.
- · Forces acting on submerged solid bodies.
- · Bouyancy and stabilitiy of floating bodies.

• Fluids in rigid body motion are also free of shear forces and their analysis is very similar to that of static fluids. They'll be studied later in ME 305.



Pressure

- For a fluid at rest, pressure is defined as the normal force acting per unit area exerted on a surface immersed in the fluid.
- It is due to the bombardment of the surface with the fluid molecules.

$1 \text{ Pa} = 1 \text{ N/m}^2$

$1 \text{ bar} = 10^5 \text{ Pa} = 100 \text{ kPa}$

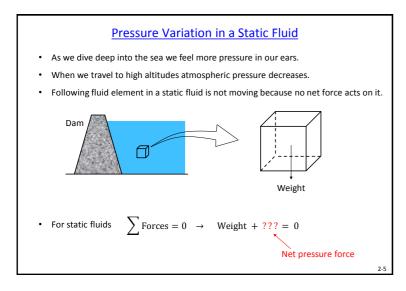
- 1 atm = 101.325 kPa = 1.01325 bars = 14.7 psi
- Atmospheric pressure that we feel is due to the air column sitting on top of us.
- It is quite high ($^{10^5}$ Newton per m² or 10 tones per m²).

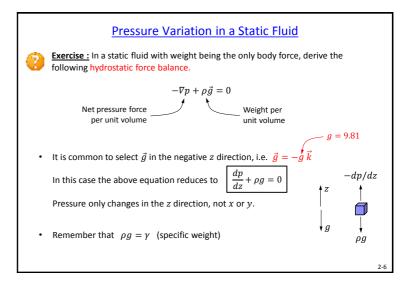


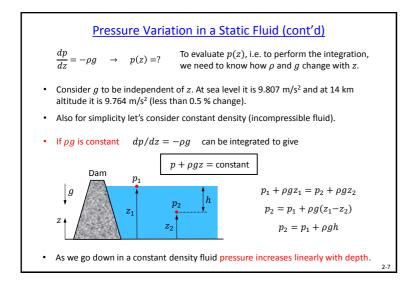
Famous Magdeburg experiment that demonstrates the power of the atmospheric nressure

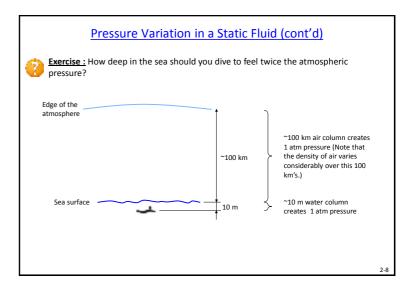
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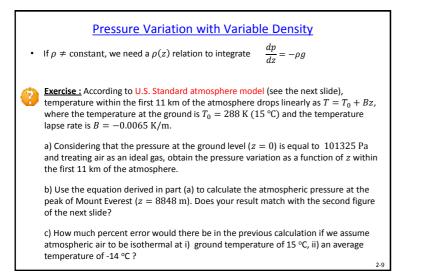
Direction Dependency of Pressure Exercise : For a fluid at rest, pressure at a point is independent of direction, which is known as Pascal's Law. Find and study its derivation in your fluid mechanics book. $p_1 = p_2 = p_3 = \cdots$ Point P Dam • In a moving fluid there will be both static and dynamic pressure definitions. Static pressure will be defined in a special way. It'll be a bit tricky. 2-4

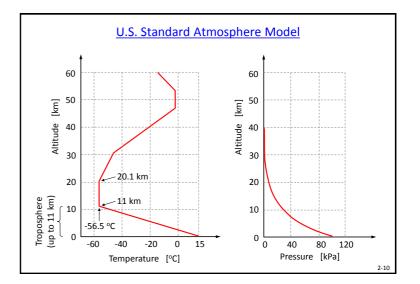












Pressure Variation with Variable Density (cont'd)

Exercise : In 1960 Trieste sea vessel carried two oceanographers to the deepest point in Earth's oceans, Challenger Deep in the Mariana Trench (10,916 m). Designers of Trieste needed to know the pressure at this depth. They performed two calculations. First they considered the seawater to be incompressible with a density equal to the value at the ocean surface, which is 1020 kg/m³. Then they considered the compressibility of seawater using a modulus of elasticity of 2.07x10⁹ Pa. Taking the atmospheric pressure at the ocean surface to be 101.3 kPa, calculate the percent error they made in the calculation of pressure at h = 10,916 m when they considered seawater to be incompressible.



Also read about James Cameron's 2012 dive at the Challenger Deep.

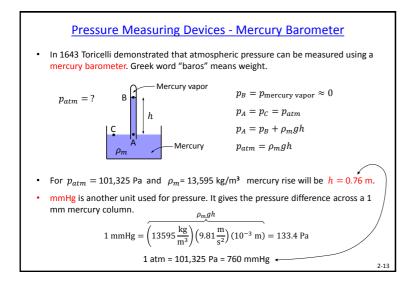
http://www.deepseachallenge.com

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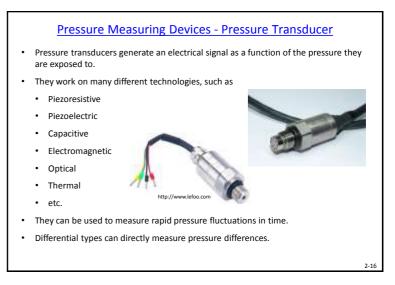
Absolute vs Gage Pressure

- Absolute pressure is measured with respect to complete vacuum.
- Certain pressure measuring devices measure pressure with respect to the ambient pressure, which is usually the atmospheric pressure. This is called gage pressure.
- Gage pressure is commonly used when we want to get rid of the atmospheric pressure effect.
- When your car's manual says that you need to inflate the tires to 30 psi, it is actually trying to say 30 psi gage (30 psi g). If the local atmospheric pressure is 95 kPa, absolute pressure of air inside the tires would be

Absolute pressure in the tire = 30 psi
$$\left(\frac{101.3 \text{ kPa}}{14.7 \text{ psi}}\right)$$
 + 95 kPa = 301 kPa
 p_{atm} = 95 kPa
Pressure gage reads
30 psi g = 206 kPa g
2.12







Pressure Measuring Devices - Bourdon Gage Measures the gage pressure. Patented at 1849. A bent elliptical tube is open and fixed at one end, and closed but free to move at the other end.

 When pressure is applied to this tube it deflects and the pointer connected to its free end shows the gage pressure (pressure with respect to the atmospheric pressure outside of the tube).

- When the tube is disconnected the pointer shows zero.
- It can be used for the measurement of liquid and gas pressures upto 100s of MPa.





Manometers

- Manometers are used to measure pressure differences using liquid columns in tubes.
- Working principles are
 - any two points at the same elevation in a continuous liquid have the same pressure.
 - pressure increases as ρgh as one goes down in a liquid column.

